BBAMEM 74506

Cation sidedness in the phosphorylation step of Na⁺/K⁺-ATPase

Harry T.W.M. van der Hijden and Jan Joep H.H.M. de Pont

Department of Biochemistry, University of Nijmegen, Nijmegen (The Netherlands)

(Received 24 February 1989)

Key words: ATPase, Na '/K '-; Cation sidedness; Phosphorylation; Liposome

Na*/K*-ATPase, reconstituted into phospholipid vesicles, has been used to study the localisation of binding sites of ligands involved in the phosphorylation reaction. Inside-out oriented Na*/K*-ATPase molecules are the only population in this system, which can be phosphorylated, as the rightside-out oriented as well as the non-incorporated enzyme molecules are inhibited by outabain. In addition, the rightside-out oriented Na*/K*-ATPase molecules have their ATP binding site intravesicularly and are thus not accessible to substrate added to the extravesicular medium. Functional binding sites for the following ligands have been demonstrated: (i) Potassium, acting at the extravellular side with high affinity (stimulating the dephosphorylation rate of the E₂P conformation) and low affinity (inducing the non-phosphorylating E₂K complex). (ii) Potassium, acting at the cytoplasmic side with both high and low affinity. The latter sites are also responsible for the formation of an E₂K complex and compete with Na* for its binding sites. (iii) Sodium at the cytoplasmic side responsible for stimulation of the phosphorylation reaction. (v) Sodium (and amine buffers) at the extracellular side enhancing the phosphorylation reaction and inhibiting it above optimal concentrations.

Introduction

In the mechanism of action of Na⁺/K⁺-ATPase the phosphorylation step plays a pivotal role [1–5]. It is assumed to be a crucial step in the conversion of chemical energy into transport. Nearly all studies on the phosphorylation reaction have been carried out in membrane preparations in which added ligands have simultaneous access to binding sites located at the extracellular and the cytosolic side of the plasma membrane. In those studies it has been established that a maximal level of phosphorylation can be obtained in the presence of both Mg²⁺ and Na⁺ and in the absence of K⁺-lons. The latter ion stimulates the rate of the dephosphorylation reaction considerably.

However, the presence of ions at both sides of the membrane precludes conclusions on the sidedness of these effects. Although it is very likely that the stimula-

Abbreviations: Tris, trighydroxynethyljaminomethane; EDTA, ethylenedinitroterracenet a end; CCCP, carbonylcyanide varieties hydrazone, $K_{\rm sort}$, Na $_{\rm sort}$ and Mg $_{\rm ger}$, represent K, Na $^{\circ}$ and Mg $^{2+}$ concentrations at the cytosolic side, respectively, $K_{\rm err}$, Na $_{\rm err}$ and Mg $_{\rm ger}$, represent these concentrations at the extraoedlular side.

Correspondence: J.J.H.H.M. de Pont, Department of Biochemistry, University of Nijmegen, P.O. Box 9101, 6500 HB Nijmegen, The Netherlands tory effects of Na⁺ on the phosphorylation reaction take place at the cytosolic and the stimulatory effects of Kt⁺ on the dephosphorylation reaction at the extracellular side, additional effects of these ions at the opposite sides cannot be excluded. Although Mg²⁺ is supposed to be involved in the binding of phosphate at the cytosolic side, possible effects at the extracellular side have not been studied before.

We therefore studied steady-state phosphorylation of Na⁺/K⁺-ATPase reconstituted in proteoliposomes, in which only phosphorylation of inside-out oriented pump molecules is possible. By variation of the ion composition in- and outside the vesicles insight has been obtained on the presence, location and function of ion-binding sites on Na⁺/K⁺-ATPase.

Materials and Methods

Preparation of Na + / K +-ATPase

Na "/K "-ĀTPase from rabbit kidney outer medulla has been prepared according to the method described by Jørgensen [6]. About 340 mg microsomes (on protein base) were incubated for one hour at 20 °C in a medium containing 0.58 mg/ml sodium oddeeyl sulfate and 25 mM imidazole-HCl (pH 7.4), 3 mM ATP and 2 mM EDTA (final protein concentration 145 mg protein per ml). After the extraction the microsomal suspension was centrifuged on a sucrose gradient (0-50%). The ATP of

the pooled fractions of the gradient was removed by incubation at 37°C in presence of Na*, Mg²* and K³ and subsequent washing. The obtained membrane fragments, enriched in Na*/K*-ATPase, were stored in indiazole buffer (25 mM, pH 7.4) containing 10% sucrose. The specific Na*/K*-ATPase activity of the preparations ranged from 1.0 to 1.6 mmol P, formed per mg protein per h.

Preparation of liposomes

Liposomes have been prepared by a reversed phase evaporation method as described by Szoka and Papa-hadjopoulos [7]. Mixtures of cholesterol, phosphatidylcholine and phosphatidylserine (ratio indicated in text and legends) in chloroform were evaporated under a stream of nitrogen to remove the organic solvent. After repeated washing with diethyl ether, a 1:1 mixture of diethyl ether and buffer solution of different composition was added and the solution thoroughly mixed on a vortex mixer, while the diethyl ether was again slowly evaporated by a stream of nitrogen. The final lipid content was between 20 and 50 mg/ml After all diethyl ether had disappeared the liposomes formed were sonicated for 30 min in a Branson sonicator bath at maximal output.

Reconstitution procedure

Purified Na+/K+-ATPase (5 mg/ml) in 20 mM imidazole buffer (pH 7.2) was partially solubilized by incubation with cholate (final concentration 0.91% w/v) during 1 min at room temperature. This mixture was then added to a 10-fold volume of liposomes, giving a lipid to protein ratio of 40 to 100 (on weight basis). After thorough mixing, the preparation was frozen in liquid nitrogen or in a mixture of dry ice and acetone and subsequently thawed at room temperature. This freezing and thawing procedure was repeated twice. Thereafter the vesicle suspension was sonicated for 6 min in a Branson sonicator bath (at maximal output). Detergent was removed from the proteoliposomes by centrifuging aliquots of the suspension over a 10-fold volume Sephadex G-25 (coarse) column (equilibrated with the appropriate buffer solution) in a syringe. This centrifugation step took 5 min (100 g) and was repeated once. More than 99.9% of the cholate was removed by this procedure [8]. Together with the removal of the detergent the extravesicular medium could be exchanged by a medium of choice [9].

ATP hydrolysis

The ATP hydrolysis was determined as the release of $^{32}P_1$ from $[\gamma^{-2}P]$ ATP [10]. To 10 μ l proteoliposomes containing Na * , Mg $^{+}$ and K * , in Tris buffer (pH 7.2), 190 μ l of a medium containing Na * , Mg $^{+}$, K * , Tris (pH 7.2), ouabain (0.2 mM) and 0.1 to 1.0 mM labeled ATP were added at room temperature. For blanc values

the reconstituted Na'/K'-ATPase was denatured with trichloroacetic acid prior to incubation with the assay medium. The $^{\rm N}P$, production was measured after stopping the reaction at a given time by addition of 0.4 ml 10% trichloroacetic acid followed by mixing with 0.4 ml 20% (w/v) aqueous charcoal suspension. The charcoal adsorbs the adenosine phosphates from the medium, but leaves $P_{\rm i}$ in solution. The suspension was mixed thoroughly during 10 s each 5 min (three times). Thereafter the charcoal was sedimented by centrifugation for 10 min at 2000 × g at 0° C. Aliquots (0.2–0.5 ml) were taken from the supernatant, mixed with 4.5 ml liquid scintillation fluid (Aqualuma Plus). Radioactivity was measured with a liquid scintillation counter.

Transport assay

For transport studies proteoliposomes loaded with 20 mM K*, were equilibrated with a medium containing "Rb* (0.1 mM) during 3 h at room temperature [11]. After loading with the tracer the proteoliposome suspension was incubated with a 10-fold volume of transport medium containing Na*, K*, Mg² and ouabain, with or without ATP. After the incubation at room temperature the transport was quenched by layering an adiquot of the suspension on a Dowes 50-X8 (Tris-form) column [12]. The proteoliposomes were then eluted with 1 ml of a solution of 250 mM sucrose and 25 mg/ml bovine serum albamin. The elutate, containing proteoliposomes devoid of the external Rb*, was counted in a liquid scintillation counter by measuring Cerenkov radiation.

Phosphory lation

Phosphorylation of the reconstituted Na*/K --ATPase was carried out at 22°C at pH 7.0. The labeled ATP concentrations varied between 0.2 and 20 µM (The Radiochemical Centre, Amersham, U.K., specific radioactivity 3000 Ci/mol). The reaction was started by rapid mixing of 10 µl proteoliposomes (preincubated with 0.2 mM ouabain and 2 mM Mg2+, if not indicated otherwise) with 90 ul of the medium containing ATP and the other ligands. The reaction was stopped after 3 s by addition of 3 ml 5% (w/v) trichloroacetic acid. containing 100 mM phosphoric acid. The denatured phosphoprotein was filtered on a 1.2 µm pore width Selectron filter (Schleicher and Schuell, Dassel, F.R.G.), which was then washed three times with 3 ml of the stopping solution. Incorporated 32 P: was determined by liquid scintillation counting. For blanc values the proteoliposomes were mixed with the stopping solution prior to addition of ATP.

Dephosphorylation

After phosphorylation for 10 s at room temperature 900 μ l of the dephosphorylation medium was added to the phosphorylation mixture (100 μ l). The dephospho-

rylation mixture contained, apart from the buffer and cations. 1 mM unlabeled ATP in order to dilute the labeled ATP 1000-fold. Together with the 10-fold dilution of the volume of the medium the final dilution of the labeled ATP was 10 000 which actually avoided further phosphorylation by the labeled ATP. After rapid mixing the dephosphorylation reaction was stopped (at the time indicated) by addition of 5 ml 5% (w/v) trichloroacetic acid containing 100 mM phosphoric acid. After stopping the reaction, the mixture was further treated as described in the phosphorylation procedure.

Materials

ATP and Tris were purchased from Boehringer, Mannheim, F.R.G. [v-3² P]ATP and ⁵⁶Rb were obtained from Amersham, Buckinghamshire, U.K., phosphatidyl-choline (egg) and phosphatidylserine (bovine brain) were purchased from Avanti Polar Lipids, Birmingham, AL, U.S.A., CCCP from Aldrich, Milwaukee, WI, U.S.A. and cholesterol, nigericin, monensin and valinomycin from Sigma, St. Louis, MO, U.S.A. All other chemicals were of reagent grade.

Results

Pump- and hydrolytic activities of the Na⁺/K⁺-ATPase containing proteoliposomes

Proteoliposomes, reconstituted by the freeze-thaw/ sonication procedure were able to extrude Rb+ actively upon addition of ATP to the medium. At 22°C the reconstituted Na*/K*-ATPase pumped 86Rb+ out of the preloaded vesicles with an initial velocity of 0.5 mmol/mg protein per h (based on total protein content). This velocity deviated from linearity after 30 s (not shown). Hydrolysis of ATP was assayed in the presence of ouabain (0.2 mM). Pi release was linear in time for about one min. Addition of the ionophores valinomycin + CCCP and nigericin increased the linearity of the P, release (Fig. 1), whereas monensin did not change the rate of hydrolysis at all. The velocity of ATP hydrolysis by the proteoliposomes was about 10 to 20% of that of the purified enzyme (at 20°C). The incorporation of the enzyme was about 50% inside-out and 50% rightside-out (determined by opening the vesicles with detergent [13]. The non-incorporated part of the enzyme population was less than 10 percent on protein base as determined by separation of the proteoliposomes from the fragmented enzyme by sucrose gradient [14]. These molecules were inactive in hydrolysis and phosphorylation because of the presence of ouabain. The rightsideout oriented Na+/K+-ATPase molecules were also silent because of the presence of the ATP site intravesiculary, inaccessible to ATP, added to the medium. Only the inside-out oriented Na*/K+-ATPase molecules could be active in the different assays used: phosphorylation, ATP-hydrolysis and active transport.

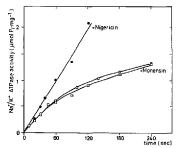


Fig. 1. ATP hydrolysis by reconstituted Na⁺/K⁺. ATPase. Na⁺/K⁻. ATPase are constituted in liposomes (phosphatidylchoine 26, phobatidylserine 2 and cholesterol 20 mg/ml) containing Tris (20 mM, pH 7.2), K⁺ (10 mM) and Na⁺ (100 mM) were preincubated with ouabain (1 mM; 60 min at room temperature). 10 μl of these proteoliposomes (protein concentration 1 mg/ml) were then incubated with 190 all of the labeled ATP-containing medium (Mgg²⁺, 1 mM; K⁺, 100 mM; Na⁻, 10 mM; ATP, 1 mM and ouabain, 0.2 mM) at room temperature and the reaction was stopped with trichloroacetic acid at the indicated time (as described in Materials and Methods). Prior to incubation the proteoliposomes were preincubated with ionobres dissolved in enhance (3 μl), to obtain a final concentration of 20 μM nigericin (Φ) or monensin (Ω). The control without ionophore (O) is also shown.

Ouabain sensitivity and phosphorylation level

The steady state level of phosphorylation of Na⁺/K⁺-ATPase containing membrane sheets was totally inhibited by ouabain ($I_{50} = 1 \mu M$), whereas only a partial decrease (< 25%) of the phosphorylation level of the Na+/K+-ATPase containing proteoliposomes was observed (Fig. 2). The decrease in phosphorylation level at low ouabain concentrations (< 10 µM) is due to the non-incorporated part of the enzyme. The resulting phosphorylation level at this ouabain concentration must be assigned to the inside-out oriented part of the pump molecules, which are not inhibited by ouabain, because they have their ouabain binding site intravesicularly. At higher ouabain concentrations a further decrease of the phosphorylation level could be seen, probably due to small amounts of ouabain diffusing into the proteoliposomes. In all further experiments the proteoliposomes were preincubated with 0.2 mM ouabain and 10 mM Mg2+ to make sure that the measured phosphorylation levels were due to inside-out oriented Na */K *-ATPase molecules. The maximal level of phosphorylation for proteoliposomes containing 200 mM Tris with 100 mM Na+, 5 mM Mg2+, 15 µM ATP and 0.2 mM ouabain in the phosphorylation medium reached 30 to 40% (based on total protein content of the

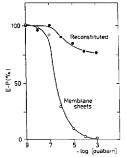


Fig. 2. Inhibition of reconstituted and fragmented Na ', X' - ATPase by ouabain. Protectiposomes with the same lipid composition as described in Fig. 1, containing Tris (200 mM, pH 72) were preincubated with the indicated concentrations of ouabain in presence of Mga* ' (2.5 mM) at room temperature. After 60 min the phosphorylation reaction was carried out at the same temperature in a medium containing Na* (100 mM), Mg² ' (2.5 mM), Tris (100 mM; pH 7.0) and labeled ATP (1 µ M) (4). Fragmented enzyme, preincubated with ouabain, was phosphorylated under the same conditions (0.1. In the absence of ouabain the steady-state phosphorylation level was 2.4 nmol/mg and 0.8 mmol/mg for the fragmented enzyme and the proteoliposomes, respectively.

proteoliposome suspension) of that of the purified enzyme under the same conditions, without outbain.

Effects on the steady-state phosphorylation level by cytosolic and extracellular sodium

In proteoliposomes containing only intravesicular buffer solution (Tris 200 mM, pH 7.0) Na_{cyt} has a

TABLE 1

Kn, Na_{syt} values for reconstituted Na+/K+-ATPase

The $K_{0.5}$ values for Na* were derived from Fig. 3a after subtraction of the Na* insensitive level by Scatchard analysis.

ATP (µM)	K _{0.5} Na _{cyt} (mM)	
1.0	12	
2.0	8.9	
5.0	6.5	
20.0	4.9	

TABLE II

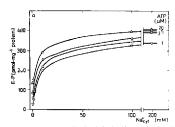
Ko., Na_{ver} values for different K_{ver} concentrations

Proteoliposomes, as described in the legend of Fig. 2, were phosphorylated in the presence of 5 mi/4 Mg²⁺ and 1 μ M ATP and K_{est} as indicated.

K _{eyt} (mM)	K _{0.5} Na _{cyt} (mM)	
0.0	10	
0.1	12	
1.0	20	
10.0	23	
200	100	

stimulating effect on the steady-state phosphorylation level. In a phosphorylation medium, containing 1 μ M ATP and 5 mM Mg²⁺, an enhancement of the phosphorylation level was according to the Michaelis-Menten formalism with a half-maximal stimulating concentration of 10.0 ± 2.5 mM (Fig. 3a). The K_{B_2} values for Na* derived from this figure are represented in Table I. The K_{D_3} value was increased by increasing K_{cyt} concentrations (Table II).

The Tris concentration at the cytoplasmic side had no effect on the steady-state phosphorylation level. In



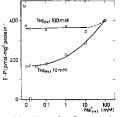


Fig. 3. Enhancement of the steady-state phosphorylation level by cytoplasmic and extracellular sodium. (a) Proteoliposomes, as described in Fig. 2 were phosphorylated in a medium containing M_g^{++} (2 mM), Tris (20 mM), B^{+} (20), incline chloride (to adjust osmolarity) and Na^{+} (as indicators of Pour different ATP concentrations were used for the phosphorylation: $1 \mu M$ (v), $2 \mu M$ (0), $5 \mu M$ (0) and 20 μM (a). (b) The same phosphorylation: 10 mM (10), and 20 μM (a), (b) The same phosphorylation: 10 mM (10), and 100 mM (0), choline chloride was added intravesicularly increasing Na^{+} content and different Tris concentration: 10 mM (10), and 100 mM (0). Choline chloride was added intravesicularly in order to keep the osmolarity constant at 380 mostM. $K_{0.5}$ values were obtained after subtraction of the Na^{+} -independent level from Scachard plots.

the absence of added Nacy, the phosphorylation level was already 10 to 20% of the maximal level (with 200 mM Nagar). Increasing the ATP concentration from 1 to 20 µM increased this level to 35% of the maximum (Fig. 3a). The phenomenon, that ATP, Mg2+ and buffer were sufficient to give a (submaximal) level of phosphorylation indicates that Na+ is not a unique cation to induce the phosphorylation. This level, which was reached in the absence of added Na+, must be due to buffer stimulated phosphorylation as described by Schuurmans Stekhoven et al. [15,16]. Extracellular sodium increased the steady-state phosphorylation level in presence of 20 mM extracellular Tris, whereas with 100 mM extracellular Tris no further increase of the phosphorylation level could be observed (Fig. 3b). With 100 mM extracellular Tris the same phosphorylation level as with 100 mM Na ext was obtained.

Cytoplasmic and extracellular Mg2+

The role of Mg³⁺ in the reaction mechanism of Na³/K⁴-ATPase is rather complex [19]. Chelating agents have to be used in studies on the role of divalent cations in the reaction mechanism of Na⁴/K⁴-ATPase and have effects themselves, which cannot easily be explained [17,18]. In Na⁴/K⁴-ATPase-containing membrane sheets a maximal phosphorylation level could be reached in the presence of 50 mM Na³ without adding any Mg²⁺ (Fig. 4). However, in the presence of 0.5 mM EDTA the steady-state phosphorylation level decreased and a dose-dependent increase in the phosphorylation level was found with a maximum between

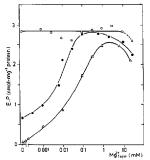


Fig. 4. Effect of Mg²⁺ and EDTA on the steady-state phosphorylation level of fragmented Na⁺/K⁺-ΛTPase. Purified enzyme was phosphorylated in a medium containing Na⁺ (50 mM). Tris (70 mM). Pt 7.0), choline chloride (30 mM). ATP (1 μM). EDTA and Mg²⁺ to get the indicated free Mg²⁺ concentration (0 mM (ο), 0.5 mM (•) or 10 mM (ο1).

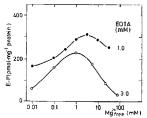


Fig. 5. Effect of cytoplasmic Mg²⁺ on the steady-state phosphorylation level in reconstituted Na⁺/K⁺-Δ-TPase. Proteoliposomes at Fig. 2, containing Tris (170 mM, pH 70) and choline chloride (30 mM) were preincubated with ouabala (0.2 mM) and Mg²⁺ C amM). After preincubation EDTA to a final concentration of 10 mM was added. Immediately thereafter the preincubated proteoliposomes were phosphorylated in a medium as described in the legand of Fig. 4. The EDTA concentration during the phosphorylation was diluted to 1 mM (Φ) and 3 mM (O). respectively.

0.1 and 1.0 mM Mg²⁺. In the presence of 10 mM EDTA the steady-state phosphorylation level was nearly zero in the absence of added Mg²⁺ and the dose-response curve was shifted to the right. At higher Mg²⁺ concentrations there was a tendency to a lowering of the E-P level.

In the reconstituted system the Mg²⁺-dependency can only be studied in the presence of EDTA, since Mg²⁺ is needed for the binding of ouabain to the non-incorporated and rightside-out oriented Na⁺/K.⁺-ATPase. Fig. 5 shows that the phosphorylation level in the reconstituted system depended on the Mg₅₊ concentration. An increase in the EDTA concentration lowered the E-P level at all Mg₅₊ concentrations and shifted the dose-response curve to the left. The Mg₅₊ concentration at which the maximal E-P level was reached was lower with higher EDTA concentrations too. In the reconstituted system the inhibitory effect at higher Mg²⁺ concentrations was much more pronounced than with the fragmented enzyme.

The Na_{cyt} dependency of the E-P level depended not only on the ATP concentration (Fig. 3a) but also on the Mg²⁺ concentration (Fig. 5). The E-P level at 1 mM Mg²⁺ was higher than at 5 mM. The effect of Mg²⁺, however, was most pronounced in absence of Na_{cyt} With 5 mM Mg²⁺ the steady-state phosphorylation level in absence of Na_{cyt} was less than 15% of the maximal level, whereas this was almost 50% with 1 mM Mg²⁺.

The affinity for ATP was influenced by cytoplasmic Mg²⁺ as well as by cytoplasmic Na⁺. An increase of the Mg²⁺ concentration increased the K_{0.5} whereas increas-

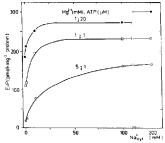


Fig. 6. Effect of cytoplasmic Mg²⁺ and ATP on the sodium dependency of the steady-state phosphorylation level. Proxelipsosmes as described in Fig. 2 were phosphorylated in the presence of Tris (20 mM, pH 7.0), choline chloride (to maintain isoosmolarity) and either 5 mM Mg²⁺ and 2 μM ATP (O). The Mg²⁺ and 2 μM ATP (O), or 1 mM Mg²⁺ and 20 μM ATP (Φ).

ing the Na^+ concentrations decreased the $K_{0.5}$ (Table III).

Modification of the extracellular Mg²⁺ concentration had no effect on the steady-state phosphorylation level (Fig. 7). The variation in these experiments is due to the fact that for changes in the extracellular concentration the intravesicular composition has to be changed so that each point represents a separate proteoliposome preparation.

Cytoplasmic potassium

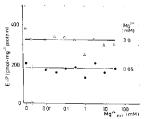
When the phosphorylation reaction was carried out with 1 μ M ATP at 0 °C with at the extracellular side 150 mM Tris only a minor inhibitory effect of K $_{cyt}$ on the steady-state phosphorylation level was observed at concentrations of K $_{cyt}$ up to 10 mM (Fig. 8). At room temperature, however, K $_{cyt}$ at low concentrations already considerably decreased the phosphorylation level.

TABLE III

ATP affinity for reconstituted Na */K *-ATPase

Proteoliposomes, as described in the legend of Fig. 2, were phosphorylated in the presence of the indicated Mg^{++} and Na^{++} concentrations with increasing ATP concentrations. $K_{0.5}$ values were derived from Scatchard plots.

Na* (mM)	Mg ²⁺ (mM)	K _{0.5} ATP (μM)
0.2	1.2	1.25
1.0	5.0	1.54
20	5.0	0.77
20	1.2	0.40
100	5.0	0.24



Increase of the ATP concentration to $10~\mu M$ or replacement of the intravesicular Tris by Na^+ decreased this potassium sensitivity. Lowering the temperature to 15° C under the latter conditions totally abolished the inhibitory effect of K^+ .

An antagonism between Na_{cyt} and K_{cyt} on the steady-state phosphorylation level could be observed under conditions where the proteoliposomes were sensitive to low concentrations of K_{cyt} . At high Na_{cyt} concentrations (200 mM) the $K_{0.5}$ for K^+ was 0.6 mM whereas at low Na_{cyt} concentrations (10 mM) the $K_{0.5}$ was only 0.95 mM (Fig. 9).

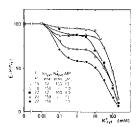


Fig. 8. Decrease of the steady-state phosphorylation level by cytoplasmic K*. Proteoliposomes (phosphatidylcholine 20 mg/ml; cholesterol 6 mg/ml), containing Tris (130 mM, pH 7.0) were phosphorylated in the presence of Na* (20 mM), Mg²* (5 mM), the indicated K_m, concentrations and 1 µM ATP at 0° C (O), at 22°C (⊕) or at 22°C with 10 µM ATP (O). Proteoliposomes with the same lipid composition, containing listidine (12 mM, pH 7.0) and Na* (100 mM) were phosphorylated with 1 µM ATP at 22°C (■) or with

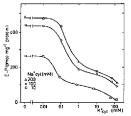


Fig. 9. Decrease of the phosphorylation level by cytosolic potassium at different cytosolic sodium concentrations. Proteoliposomes as described in the legend of Fig. 2 were phosphorylated in the presence of Mg2+ (5 mM). ATP (1 μM) and Na_{cyt} 10 mM (□), 100 mM (○) or 200 mM (a), with the K cyi concentrations as indicated.

Under all conditions mentioned above a decrease in the steady-state phosphorylation level was observed at Keyt concentrations above 10 mM. The I50 value for the latter effect was between 60 and 100 mM. This decrease is probably due to occupation of the Na+ sites by Keyl yielding an occluded E2K conformation [20,21]. The decrease of the steady-state phosphorylation level was not due to an increase of the dephosphorylation rate, because the dephosphorylation rate in the presence of Key was not significantly higher than in absence of this ligand (not showr.).

Extracellular potassium

Fig. 10 shows, that Kext reduced the steady-state phosphorylation level only for 15% at concentrations

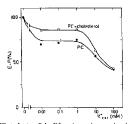


Fig. 10. Effect of extracellular K + on the steady-state phosphorylation level. Proteoliposomes with (O) and without (O) cholesterol (20 mg/ml, phosphatidylserine was 4 mg/ml and phosphatidylcholine 26 and 46 mg/ml, respectively) were prepared with different concentrations of extracellular K+. The proteoliposomes contained additionally Tris (100 mM; pH 7.0) and choline chloride to maintain isoosmolarity.

The phosphorylation medium was the same as in Fig. 2.

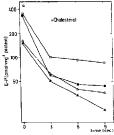
between 0.01 and 1 mM. At higher concentrations a further decrease in the E-P level was observed. The small decrease at low Kest concentrations is unexpected since in fragmented enzyme preparations a nearly completely decrease in E-P is found. An explanation could be that in the reconstituted system the amount of E, P is very small. Yoda and Yoda [22,23] have suggeted that part of the phosphorylated intermediate is in the E*P form (already proposed by Nørby and Kłodos [28,29]) which only can be deposphorylated by Kext after conversion into E_2P . The ratio of E_2P to $E^*P + E_1P$ has reported to be increased with decreasing cholesterol content of Na+/K+-ATPase-containing liposomes. The above experiment was therefore repeated with proteoliposomes containing no cholesterol, whereas in all other experiments the cholesterol content was 40%. At Key concentrations between 0.01 and 1 mM there was a significant further decrease in the steady-state phosphorylation level compared to that in cholesterol containing proteoliposomes. At Kext concentrations above 10 mM the E-P level further decreased as was the case for the cholesterol containing proteoliposomes. An explanation for the latter inhibition could be the formation of an E1K conformation which cannot be phosphorylated anymore.

Dephosphorylation

Since steady-state phosphorylation levels are the result of phosphorylation and dephosphorylation reactions, interpretation of the effects of added K+ on the E-P level is complex. Therefore the phosphorylated intermediate was first prepared during 10 s, whereafter the radioactive ATP was diluted 10000-fold. The E-P level was measured at various time points. Dephosphorylation curves showed non-linear behaviour in a semilogarithmic plot. Fig. 11 indicates that the dephosphorylation rate in the first 3 s was faster than in the following 6 s. The steady-state phosphorylation level in proteoliposomes containing a high content of cholesterol (Fig. 11a) was only 5% reduced when 10 mM Kext was present in the vesicles either at 37°C or at 0°C. At 37°C the steady-state phosphorylation level was about half of that at 0°C. The dephosphorylation rate in the first 3 s was, however, faster in vesicles containing K than in vesicles without Kext. The relative increase of the dephosphorylation rate was larger at 0°C than at

The proteoliposomes without cholesterol (Fig. 11b) showed a larger difference in the steady-state phosphorylation level at 0°C between vesicles with and without Kext (34%). The fast dephosphorylation rate in the first 3 s was for both preparations the same.

Addition of 1 mM ADP also increased the rate of dephosphorylation suggesting that part of the phosphorylated intermediate was ADP-sensitive. When the phosphointermediate was formed in the presence of



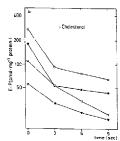


Fig. 11. Dephosphorylation of reconstituted Na⁷/K*-ATPase. (a) Proteoliposomes (as in Fig. 8) containing Tris (12 mM, pH 7.0), Na* (50 mM) and choline chloride (100 mM) (open symbols) or Tris (12 mM, pH 7.0), Na* (50 mM), choline chloride (90 mM) and K* (10 mM) (closed symbols) were phosphorylated urining 10 s and dephosphorylated during the indicated time. Reactions were carried at 0° C (squares) and 37° C (triangles). Proteoliposomes containing no Na* showed an identical dephosphorylation pattern. (b) The same dephosphorylation experiment was carried out with proteoliposomes without cholesterol (phosphatidylcholine was 26 mg/ml) loaded with Tris (200 mM, pH 7.0) and 37° C by triangles.

Tris (190 mM, pH 7.0) and K* (10 mM) (closed symbols). Dephosphorylation at 0° C is represented by circle at 37° C by triangles.

K_{eyt}, the ADP-sensitive fraction was reduced to very low levels (not shown).

Discussion

Proteoliposomes as tools for sidedness studies

Na+/K+-ATPase reconstituted in lipid vesicles provides a good system to study such properties as transport and sidedness of ligand activation or inhibition on transport and overall or partial reactions. The composition of the intravesicular medium is totally under control and can be chosen during preparation of the proteoliposomes. Because it is possible to separate (or exchange) the extravesicular medium by gel filtration or ion exchange chromatography for any medium of choice, control of the extravesicular medium is also guaranteed. In studying a fast process as phosphorylation, where only short periods of incubation with ligands are necessary, the hazard of leakage of ligands in either direction is almost excluded. In the present study proteoliposomes were always preincubated with ouabain and Mg2+ at a concentration which inhibits more than 97% of non-incorporated ATPase molecules. The rightsideout oriented Na+/K+-ATPase has its ATP binding site inside the vesicle, which cannot be reached by externally added ATP. This leaves only a single population: the inside-out incorporated pump molecules as subject for our phosphorylation studies. The orientation of the latter pump molecules makes the cytoplasmic side accessible from the medium: hence variation in the composition of the extravesicular medium represents a variation at the cytoplasmic side of these ATPase molecules.

Na + and buffer cations

Titrating the Na+ concentration in the phosphorylation medium (Nacyt), which also contained Mg2+ and ATP, increased the phosphorylation level in a similar way as with fragmented Na+/K+-ATPase. The mechanism of this activation is probably the formation of the E, conformational state of the enzyme (Fig. 12), which can bind ATP and subsequently can be phosphorylated [24]. The maximal level of phosphorylation, however, was only attained when the extracellular (intravesicular) medium contained a minimal concentration of Tris. The latter buffer could be successfully replaced by sodium or by other buffers like imidazole, triethanolamine, triallylamine or histidine, but not by choline chloride or sorbitol. This indicates that the presence of either sodium or one fo the amine buffers at the extracellular side is a prerequisite for high phosphorylation levels. It im-

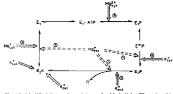


Fig. 12. Modified Post-Albers Scheme for Na'/K'-ATPase in which the sidedness of ligands is given. (+) indicates a stimulatory and (-) an inhibitory effect on the transitions (in a clockwise direction of the reaction cycle). = π=> indicate the three possible sites of effects in the reaction cycle as discussed in the text.

plicitly proves the presence of extracellular binding sites for Na+ and amine buffers. The mechanism of this extracellular effect of Na+ or amine buffers is not clear but could be due to inhibition of the dephosphorylation rate (Fig. 12) as described by Schuurmans and Stekhoven et al. [25] and by Fukushima [26]. Retardation of the dephosphorylation reaction would lead to accumulation of phosphorylated intermediates. An alternative explanation is that the enzyme is partly in the E₂ conformation and that extracellular ligands induce an E, conformation, which can bind ATP and can subsequently be phosphorylated (Fig. 12). Such allosteric effect of Na+ has been reported by Karlish and Stein [27]. A third explanation is that the extracellular ligands induce a shift to the left in the E1P + E2P equilibrium (Fig. 12). Such a shift would in itself not have an effect on the phosphorylation level, but if the hydrolysis of E₂P during the three seconds of the phosphorylation period is faster than that of E1P, it would lead to an increase in the steady-state phosphorylation level [33]. Further study is required to establish which of these three alternatives is valid.

The effects of K +

Extracellular K+ is supposed to reduce the steadystate phosphorylation level by increasing the dephosphorylation rate (Fig. 12). However, under the standard conditions mentioned above Kext reduces the steadystate phosphorylation level in proteoliposomes containing high cholesterol by only 15%. According to the third explanation mentioned above the reduction of the phosphorylation level by Kext occurs through the E2P part of the phosphorylated intermediates. This suggests that in the proteoliposomes containing high levels of cholesterol most of the phosphorylated intermediates were either in the E,P or the E*P form. The latter form is postulated by Nørby and Klodos [28,29]. In proteoliposomes containing no cholesterol low concentrations of K were able to lower the steady-state phosphorylation level by 30% indicating that the amount of E2P was increased. This finding is in agreement with the observation of Yoda and Yoda [23] that cholesterol shifts the E*P ↔ E2P equilibrium to the left. The reduction of the steady-state phosphorylation level at high concentrations of Kext is consistent with results of Karlish and Stein [31]. Blostein and Chu [32] however, found a total reduction by 100 µM Key, of the steady-state phosphorylation level in erythrocyte ghosts without Na ext and no reduction in ghosts with Na ext. This might indicate, that our reconstituted system with high intravesicular buffer concentrations represents a system with properties in between the two extremes of the ghost preparations with and without Naest. Possibly the enzyme is mainly in an E+P form and only a minor part in the E,P form. A similar suggestion was made by Yoda and Yoda [33].

Both in cholesterol-containing and in cholesterol-free proteoliposomes K* concentrations above 1 mM gave a further decrease in the steady-state phosphorylation level. The high K* concentration needed for this effect makes it unlikely that the effect is due to an increase in the dephosphorylation rate. We suggest that an E₂K complex was formed which could not be phosphorylated anymore (Fig. 12).

Reduction of the steady-state phosphorylation level by low K cyt concentrations could only be observed under special conditions. A high-affilty K+ binding site is proposed by other authors [27,34,35]. Presence of 150 mM Tris, absence of Na ext, los ATP concentration and an incubation temperature of 22°C resulted in a 40-50% reduction in the steady-state phosphorylation level at 1 mM Kcyi. Increase in the ATP or Nacxt concentration (or a decrease in the Tris concentration) and a lowering of the temperature led to a diminishment of this effect. This might be the reason, why others [23,32] observed that the phosphorylation in inside-out oriented Na⁺/K⁺-ATPase in proteoliposomes and erythrocytes was insensitive to Keet. Under the given conditions (high extracellular buffer concentration and 22°C the dephosphorylation rate of E₂P is considered to be high, but at the same time the intravesicular buffer blocks the transformational change from E*P to E2P [30]. These conditions make it possible to obtain a shift of E*P to E2P. The blockade of this conformational change by extracellular Tris was apparently overruled by Keyt (Fig. 12) and a decrease in the phosphorylation level was observed as a consequence of the fast dephosphorylation. With low extracellular Tris or histidine concentrations the E*P to E2P conformational change was apparently not inhibited, but the dephosphorylation reaction was slow. As a consequence the change of E*P to E₂P could not be observed as a decrease in the steadystate phosphorylation level. At 0°C Kcvt apparently was not able to induce the shift between E*P and E_P and therefore no decrease in the steady-state phosphorylation level could be observed. The latter result is in accordance with that of Blostein and Chu [32].

At higher $K_{\rm cyt}$ concentrations (10–200 mM) a decrease of the phosphorylation level was observed under all conditions mentioned above, as was the case with high $K_{\rm cyt}$ concentrations. This reduction of the level can be explained by occupation of the low-affinity K^* -sites. Occupation of these sites by K^* prevents the conversion of E_2 to E_1 which means that part of the enzyme cannot be phosphorylated anymore [31]. This is confirmed by the antagonism between $Na_{\rm cyt}$ and ATP on the one and $K_{\rm cyt}$ on the other hand. The increase by K^* of the dephosphorylation rate of fragmented enzyme is in the light of our model the result of two simultaneous effects on the enzyme: (1) the shift of E_1P and E^*P towards E_2P , induced by $K_{\rm cyt}$ (which is probably very fast) and (2) the increase of the dephosphorylation rate of E_2P by

K_{ent}. A remarkable discrepancy between membrane sheets and reconstituted enzyme was found. In presence of 10 mM K⁺ and 100 mM Na⁺ the steady-state phosphorylation level of Na⁺/K⁺-ATPase in fragmented membrane sheets was practically zero [15]. With 10 mM K⁺ and 100 mM Na⁺ at both sides of the reconstituted enzyme we found phosphorylation levels which reached 26% of the maximum in absence of K⁺ under otherwise the same conditions. The results obtained with the reconstituted enzyme, however, establish the presence of K⁺-binding sites both at the extracellular and at the intracellular side of the membrane.

Effects of Mg2+

Variation of the free extracellular Mg2+ concentration has no influence on the steady-state phosphorylation level. The cytoplasmic Mg2+ concentration however increases the steady-state phosphorylation level at low Mg2+ concentrations and lowers it at high Mg2+ concentrations. The magnitude of the effect depends similar to that in the fragmented enzyme on the EDTA concentration. This effect of EDTA, which can be observed even more clearly in fragmented enzyme [17,28]. is not easy to explain. Possible explanations are direct inhibitory effects of EDTA, complexation of unknown divalent cations which function can be imitated by Mg2+ or a change in the lipid-protein interaction which can only be reversed by high Mg2+ concentrations. In the reconstituted enzyme the inhibitory effect of high Mg2+ concentrations is more pronounced than in the fragmented enzyme. In the proteoliposomes Mg2+ stimulates the phosphorylation in the presence of Na_{cut} and low ATP concentrations. In the absence of Na eve however high levels of phosphorylation can only be obtained when the Mg2+ concentration is below 1 mM. Higher concentrations of Mg2+ strongly decrease the phosphorylation level. This antagonism has also been observed for the buffer stimulated phosphorylation of fragmented enzyme [15]. In the absence of Na ext and at relatively high ATP concentration (20 µM) the reconstituted enzyme is phosphorylated to 80% of the maximal level attained in presence of saturating Na., This level is obtained with extracellular imidazole, but also with (high concentrations of) Tris. With fragmented enzyme, where ligands can approach the enzyme from both sides. Tris does not stimulate phosphorylation in the absence of Na+. This different behaviour of Tris can be explained by its inhibitory capacity at the cytoplasmic side, or by the different lipid environment of the reconstituted enzyme. The stimulating effect of buffers is only observed at the extracellular side, whereas Na+ has a dual effect by enhancing phosphorylation from the cytoplasmic as well as from the extracellular side.

Conclusions

In conclusion our findings indicate that: (1) High buffer (or Na1) concentrations are necessary at the extracellular side to obtain maximal phosphorylation levels. (2) Amine buffer-stimulated phosphorylation is suppressed by Mg2+ and low ATP concentration. (3) Cytoplasmic K * reduces the steady-state phosphorylation level at low concentrations by shifting the EP intermediates towards the E-P conformation and at high concentrations via occupation of the low-affinity K+-sites by yielding the K+-occluded state of the enzyme. (4) Extracellular K' reduces the steady-state phosphorylation level partially, i.e. only those EP intermediates which are in the E₂P conformation will be hydrolysed. (5) Binding sites are demonstrated for: Kox (high and low affinity), Key (high and low affinity), Na Na Mg2+ (cytoplasmic) and amine buffers at the extracellular side.

Acknowledgements

The authors would like to thank Dr. F.M.A.H. Schuurmans Stekhoven for his useful comments on this manuscript. This investigation was financially supported by the Netherlands Organization for Scientific Research (N.W.O.), through the Netherlands Biophysics Foundation.

References

- 1 Glynn, I.M. and Karlish, S.J.D. (1975) Annu. Rev. Physiol. 37, 13-55.
- 2 Schuurmans Stekhoven, F.M.A.H. and Bonting, S.L. (1981) in Membrane Transport, New Comprehensive Biochemistry, Vol. 2 (Bonting, S.L. and De Pont, J.J.H.H.M., eds.), pp. 159–182, Elsevier-North Holland Biomedical Press, Amsterdam.
- 3 Jørgensen, P.L. (1982) Biochim, Biophys. Acta 694, 27-68.
- 4 Robinson, J.D. (1933) in Current Topics in Membranes and Transport, Vol. 19 (Bronner, F., Kleinzeller, A., Hoffman, J.F. and Forbush, B. III, eds.), pp. 485-512, Acadamic Press, New York.
- 5 Skou, J.C., Norby, J.G., Maunsbach, A.B. and Esmann, M., eds. (1988) The Na*, K*-pump Part A: Molecular Aspects, Alan R. Liss, New York.
- 6 Jørgensen, P.L. (1974) Biochim. Biophys. Acta 356, 36-52.
- 7 Szoka, F. and Papahadjopoulos, D. (1978) Proc. Natl. Acad. Sci. USA 75, 4194–4198.
- 8 Skrabanja, A.T.P., Asty, P., Soumarmon, A., De Pont, J.J.H.H.M. and Lewin, M.J.M. (1986) Biochim. Biophys. Acta 860, 131–136.
- 9 Penefsky, H.S. (1977) J. Biol. Chem. 252, 2891–2899.
- 10 Fu, Y.F., Schuurmans Stekhoven, F.M.A.H., Swarts, H.G.P., De Pont, J.J.H.H.M. and Bonting, S.L. (1985) Biochim. Biophys. Acta 817, 7-16.
- 11 Karlish, S.J.D. and Pick, U. (1981) J. Physiol. 312, 505-529.
- 12 Gasko, O.D., Knowles, A.F., Shertzer, H.G., Soulinna, E.M. and Racker, E. (1976) Anal. Biochemistry 72, 57-65.
- Cornelius, F. and Skou, J.C. (1984) Biochim. Biophys. Acta 772, 357–373.
- 14 Skrabanja, A.T.P., Van der Hijden, H.T.W.M. and De Pont, J.J.H.H.M. (1987) Biochim. Biophys. Acta 903, 434–440.
- 15 Schuurmans Stekhoven, F.M.A.H., Swarts, H.G.P., De Pont. J.J.H.H.M. and Bonting, S.L. (1985) Biochim. Biophys. Acta 815.

- 16-24.
- 16 Schuurmans Stekhoven, F.M.A.H., Swarts, H.G.P., 't Lam, G.K., Zou, Y.S. and De Pont, J.J.H.H.M. (1988) Biochim. Biophys. Acta 937 161-176
- 17 Klodos, I. and Skou, J.C. (1975) Biochim. Biophys. Acta 391, 474-485.
- 18 Klodos, I. and Skou, J.C. (1977) Biochim. Biophys. Acta 481, 667-679.
- 19 Sachs, J.R. (1988) J. Physiol. 400, 575-591.
- 20 Post. R.L., Hegyvary, C. and Kume, S. (1972) J. Biol. Chem 247, 6530-6540.
- 21 Forbush, B. III (1987) J. Biol. Chem. 262, 11104-11115.
- 22 Yoda, S. and Yoda, A. (1987) J. Biol. Chem. 262, 103-109.
- 23 Yoda, S. and Yoda, A. (1988) J. Biol. Chem. 263, 10320-10325.
- 24 Rephaeli, A., Richards, D. and Karlish, S.J.D. (1986) J. Biol.
- Chem. 261, 6248-6254.

- 25 Schuurmans Stekhoven, F.M.A.H., Swarts, H.G.P., De Pont, J.J.H.H.M. and Bonting, S.L. (1986) Biochim. Biophys. Acta 855, 375-382.
- 26 Fukushima, Y. (1987) J. Biol. Chem. 262, 11000-11005.
- 27 Karlish, S.J.D. and Stein, W.D. (1985) J. Physiol. 359, 119-149.
- 28 Nørby, J.G., Klodos, I. and Christiansen, N.O. (1983) J. Gen. Physiol. 82, 725-759.
- 29 Klodos, I. and Nørby, J.G. (1987) Biochim. Biophys. Acta 897, 302-314.
- 30 Yoda, S. and Yoda, A. (1986) J. Biol. Chem. 261, 1147-1152. 31 Karlish, S.J.D. and Stein, W.D. (1982) J. Physiol. 328, 295-316.
- 32 Blostein, R. and Chu, L. (1977) J. Biol. Chem. 252, 3035-3043.
- 33 Yoda, A. and Yoda, S. (1987) J. Biol. Chem. 262, 110-115.
- 34 Matsui, H. and Homareda, H. (1982) J. Biochem, 92, 193-217.
- 35 Jørgensen, P.L. and Petersen, T. (1982) Biochim. Biophys. Acta 705, 38-47.